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Chemistry If8766 . MOLARITY BY DILUTION Name Acids are usually acquired from chemical supply houses in concentrated form. These acids are diluted to the desired concentration by adding water. Since moles of acid before dilution = moles of acid after dilution, and moles of acid =  $M \times V$  then,

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Chemistry If8766 Instructional Fair Inc - WordPress.com Molarity is expressed in units of moles per liter (mol/L). It's such a common unit, it has its own symbol, which is a capital letter M. A solution that has the concentration 5 mol/L would be called a 5 M solution or said to have a concentration value of 5 molar.

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where the subscripts "1" and "2" refer to the solution before and after the dilution, respectively. Since the dilution process does not change the amount of solute in the solution,  $n_1 = n_2$ . Thus, these two equations may be set equal to one another:  
$$[M_1 V_1 = M_2 V_2]$$
 This relation is commonly referred to as the dilution equation.

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tool calculates the volume of stock concentrate to add to achieve a specified volume and concentration. The calculator uses the formula  $M_1 V_1 = M_2 V_2$  where "1" represents the concentrated conditions (i.e. stock solution Molarity and volume) and "2" represents the diluted ...

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Molarity is expressed in units of moles per liter (mol/L). It's such a common unit, it has its own symbol, which is a capital letter M. A solution that has the concentration 5 mol/L would be called a 5 M solution or said to have a concentration value of 5 molar.

## **Molarity Definition as Used in Chemistry**

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